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## A Heterotetranuclear [NillRe<sup>IV</sup><sub>3</sub>] Single-Molecule Magnet

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Since the discovery of discrete species which can exhibit the phenomenon of single-molecule magnetism,<sup>1</sup> a large amount of work has been devoted to their preparation and characterization because of the need for knowledge of their fundamental properties [systems where exists an energy barrier to the relaxation of the magnetization which is due to the occurrence of a large ground spin state with a significant negative zero-field splitting (D)] and potential applications (molecular units for data storage and quantum computation).<sup>2,3</sup> The majority of the examples of single-molecule magnets (SMMs) concerns homometallic Mn polynuclear complexes where the anisotropy is created by the presence of Jahn-Teller distorted Mn<sup>III</sup> ions. The need for a larger magnetic anisotropy, higher ground spin states and a better understanding of quantum tunneling effects through synergy of heterometal spins, moved preparative chemists toward the heterometallic SMMs as illustrated by recent reports on mixed 3d-4f complexes.<sup>4</sup>

In the context of our research activity in molecular magnetism, we have focused on the design stable mononuclear Re(IV) complexes and their use as ligands toward first row transition metal ions.<sup>5</sup> Re(IV) is a 5d<sup>3</sup> ion which forms usually octahedral complexes that have two very important characteristics from a magnetic viewpoint: (i) a large degree of spin delocalization on the ligands in its complexes because of covalency effects,<sup>6,7</sup> and (ii) its remarkable magnetic anisotropy which is due to the high value of the spin-orbit coupling parameter ( $\lambda$  ca. 1000 cm<sup>-1</sup> in the free ion).5 These features together with the well-known capability of the oxalate (ox) ligand to transmit electronic interactions between the magnetic centers when acting as a bridge,<sup>8</sup> make the mononuclear complex [ReCl<sub>4</sub>(ox)]<sup>2-</sup> a very suitable candidate for metal assembling in magneto-structural studies.

In this work we present the first example of a tetranuclear Ni<sup>II</sup>- $\operatorname{Re}^{\operatorname{IV}_3}$  complex of formula  $(\operatorname{NBu}_4)_4[\operatorname{Ni}{\operatorname{ReCl}_4(\operatorname{ox})}_3]$  (1) which behaves as a SMM that is prepared by the reaction of nickel(II) nitrate hexahydrate and (NBu<sub>4</sub>)<sub>2</sub>[ReCl<sub>4</sub>(ox)] in nonaqueous solutions.

The structure of 1 consists of discrete tetranuclear  $[Ni{ReCl_4(ox)}_3]^{4-}$ anions (Figure 1) and  $[n-(C_4H_9)_4N]^+$  cations which are held together by electrostatic forces. Both metal atoms are six coordinated: six oxalate-oxygen atoms (at the Ni) and four chloro and two oxalateoxygen atoms (at the Re) build somewhat distorted octahedral surroundings. The values of the Ni-O, Re-O, and Re-Cl bond lengths vary in the ranges 2.059(4)-2.095(4), 2.058(5)-2.097(4), and 2.283(3)-2.347(2) Å, respectively. The values of the bite angle subtended by the oxalate cover the ranges  $79.2(2)-79.7(2)^{\circ}$  (at the Re) and  $81.6(2)-82.4(2)^{\circ}$  (at the Ni). The values of the dihedral angles between the mean planes of the oxalate ligands are 88.4(1),

89.0(1), and 77.3(1)°. The degree of twist of the trischelated environment of the nickel atom is 49° (the twist angle  $\phi$  is 60° for a regular octahedron).<sup>9</sup> The four metal atoms of the tetranuclear unit are coplanar and each polynuclear entity is well separated from each other (Figure S1) [the shortest intermolecular Re-Re and Cl-Cl separations being 9.803(1) and 5.608(3) Å, respectively]. The average Ni-Re distance across the bridging oxalate is 5.387(1) Å.

The magnetic properties of **1** under the form of  $\chi_{\rm M}T$  versus T plot ( $\chi_M$  is the magnetic susceptibility per Ni<sup>II</sup>Re<sup>IV</sup><sub>3</sub> unit) are shown in Figure 2.  $\chi_M T$  at room temperature is 6.73 cm<sup>3</sup> mol<sup>-1</sup> K, a value which is as expected for one nickel(II) ( $S_{Ni} = 1$ ) and three rhenium-(IV) ( $S_{\text{Re}} = \frac{3}{2}$ ) ions magnetically isolated. Upon cooling,  $\chi_{\text{M}}T$ continuously increases to reach a maximum value of 15.8 cm3 mol-1 K at 6.0 K and further decreases to 14.4 cm<sup>3</sup> mol<sup>-1</sup> K at 1.9 K. These features are typical of a significant intramolecular ferromagnetic interaction [a case of strict orthogonality between the interacting magnetic orbitals ( $t_{2g}$  vs  $e_g$ )] with a low-lying S = 11/2spin state ( $\chi_{\rm M}T = 16.65 \text{ cm}^3 \text{ mol}^{-1} \text{ K}$  for  $g_{11/2} = 1.93$ ).<sup>10</sup> The small decrease of  $\chi_{\rm M}T$  at very low temperatures is most likely due to zero-field splitting effects. The magnetic data were fitted through the Hamiltonian of eq (1) where the first term is the isotropic exchange interaction

$$\hat{\mathbf{H}} = \hat{\mathbf{H}}_{\text{exchange}} + \hat{\mathbf{H}}_{\text{ZFS}} + \hat{\mathbf{H}}_{\text{Zeeman}}$$

$$\hat{\mathbf{H}}_{\text{exchange}} = -J_{\text{ReNi}}\hat{S}_{\text{Ni}}\sum_{i=1}^{3}\hat{S}_{\text{Rei}}$$

$$\hat{\mathbf{H}}_{\text{ZFS}} = D_{\text{Re}}[\sum_{i=1}^{3}\hat{S}_{\text{ZRei}}^2 - \mathbf{15/4}] + D_{\text{Ni}}[\hat{S}_{\text{ZNi}}^2 - \mathbf{2/3}]$$

$$\hat{\mathbf{H}}_{\text{Zeeman}} = g_{\text{Re}}^{\text{H}}\beta H\sum_{i=1}^{3}\hat{S}_{\text{ZRei}} + g_{\text{Ni}}^{\text{H}}\beta H\hat{S}_{\text{ZNi}} + g_{\text{Re}}^{\perp}\beta H$$

$$[\sum_{i=1}^{3}\hat{S}_{\text{XRei}} + \sum_{i=1}^{3}\hat{S}_{\text{YRei}}] + g_{\text{Ni}}^{\perp}\beta H(\hat{S}_{\text{XNi}} + \hat{S}_{\text{YNi}}) \quad (1)$$

between the Re<sup>IV</sup> and Ni<sup>II</sup> ions, the second one is the zero-field splitting of the interacting cations, and the last term accounts for the Zeeman effects. In order to avoid overparametrization, we assumed that all the rhenium centers are equivalent and  $g_{\parallel} = g_{\perp}$ for both Re<sup>IV</sup> and Ni<sup>II</sup> ions. Given the strong correlation observed in the fitting procedure between the  $D_{\text{Re}}$  and  $D_{\text{Ni}}$  parameters, we made  $D_{\rm Ni} = 0$ . The values of the best-fit parameters were J =+16.3(3) cm<sup>-1</sup>,  $g_{\text{Ni}} = 2.18(1)$ ,  $g_{\text{Re}} = 1.87(1)$ , and  $|D_{\text{Re}}| = 2.8(2)$  $cm^{-1}$ . Assuming that the local D tensors are collinear, a value of  $D_{11/2} = 0.46 \text{ cm}^{-1}$  is calculated through the expression  $D_{11/2} =$ 0.163 D<sub>Re</sub>

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**Figure 1.** ORTEP drawing of the tetranuclear  $[Ni{ReCl_4(ox)}_3]^{4-}$  anion of 1 with thermal ellipsoids at the 30% probability level.



Figure 2. Thermal variation of the  $\chi_M T$  product for 1 under an applied magnetic field of 100 G: (o) experimental data; (-) best-fit curve (see text). The inset shows the frequency and temperature dependence of the out-of-phase ac signal under external applied dc magnetic fields of 2000 (top) and 0 G (bottom) in a 1 G oscillating field and in the frequency range 100-1400 Hz.

 $+ 0.0182 D_{\rm Ni}$ . These results show that the only populated spin state at low temperatures is the S = 11/2 ground spin state. The small value computed for  $D_{\rm Re}$  contrasts with the greater one for this parameter in the mononuclear [ReCl<sub>4</sub>(ox)]<sup>2-</sup> anion.<sup>5</sup> A non collinear situation of the local D tensors in 1 would account for this difference. The computed value of  $D_{11/2}$  agrees with that extracted from the magnetization plot at 2.0 K ( $D_{11/2} = -0.50(5)$  cm<sup>-1</sup>, Figure S2)

Frequency dependent out-of-phase ac signals are observed [inset of Figure 2 and Figure S3 (left)] which are indicative of a system with slow relaxation of the magnetization. For H = 0, no maxima of  $\chi''_{M}$  are observed above 1.9 K in the frequency range explored. However, maxima of  $\chi''_{\rm M}$  are observed at T > 1.9 K for H = 500, 1000, and 2000 G, their position being shifted to greater temperatures with the increasing field. The data obtained were fit to the Arrhenius equation to give values of  $\tau_0$  (preexponential factor) and  $E_{\rm a}$  (energy barrier) of 8.5 × 10<sup>-9</sup> s and 16.2 cm<sup>-1</sup> (H = 2000 G),  $2.3 \times 10^{-9}$  s and 14.2 cm<sup>-1</sup> (H = 1000 G), and 2.9 × 10<sup>-8</sup> s and

13.2 cm<sup>-1</sup> (H = 500 G) (Figure S3, right). By extrapolation from these data, one can see that the value of  $\tau_0$  for H = 0 G must be close to  $3 \times 10^{-8}$  s. Also the linear field dependence of the energy barrier  $E_a = U^{\#} + MH$  occurs with M = 0.002 cm<sup>-1</sup> G which leads to  $U^{\#} = 12.2 \text{ cm}^{-1}$  for H = 0 G [Figure S3, right (bottom)]. As in the present case  $U^{\#} = D_{11/2}(S^2 - \frac{1}{4})$  with S = 11/2, a value of  $0.41 \text{ cm}^{-1}$  is derived for  $D_{11/2}$  in agreement with the value obtained for this parameter through the above fit of the  $\chi_{\rm M}T$  data.

In order to investigate possible SMM behavior, single-crystal magnetization measurements were performed with an array of micro-SQUIDS at temperatures down to 40 mK. Hysteresis loops for 1 at varying temperatures and sweep rates are shown in Figure S4. These curves show that an extremely fast tunneling occurs at zero applied field. However, when a field is applied the tunneling is switched off and sweep rate and temperature-dependent hysteresis loops occur, a behavior which was previously observed in cyanobridged heterometallic species.<sup>11</sup>

In summary, 1 is best described as either a SMM with a fast tunnel transition at H = 0 G or as a compound whose SMM behavior is only "switched-on" by the application of an external magnetic field.

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Supporting Information Available: Figures S1-S4, experimental details, and the CIF file of 1. This material is available free of charge via the Internet at http://pubs.acs.org.

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